

## ISOFLAVONES OF TWO *IRIS* SPECIES

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**Key Word Index**—*Iris milesu*, *Iris kumaonensis*, Iridaceae, rhizomes, isoflavones

**Abstract**—Analysis of the methanol extracts of the rhizomes of *Iris milesu* resulted in the isolation of a new isoflavone, 5,6,7,4'-tetrahydroxy-3'-methoxyisoflavone and that of *Iris kumaonensis*, iriskumaonin methyl ether, iriskumaonin, irisflorentin, junipegeenin-A, irigenin and iridin

### INTRODUCTION

Recently we reported a new isoflavone from *Iris milesu* [1]. Further work on the same species has now afforded another new isoflavone characterized as 5,6,7,4'-tetrahydroxy-3'-methoxyisoflavone. *Iris kumaonensis* on chemical investigation resulted in the isolation of iriskumaonin methyl ether [2], irisflorentin [3, 4], junipegeenin-A [5], being reported here for the first time from *Iris kumaonensis*, along with iriskumaonin [6], irigenin and iridin. Junipegeenin-A has been reported recently only from *Juniperus macropoda* (Cupressaceae).

### RESULTS AND DISCUSSION

Compound 1, mp 257°, analysed for  $C_{16}H_{12}O_7$  ( $M^+$  316) and gave a strong  $FeCl_3$  test. Its IR spectrum displayed bands at 3600–3200 and 1650  $cm^{-1}$  suggesting the presence of a chelated C=O group. It formed a tetraacetate, mp 88°, analysing for  $C_{24}H_{20}O_{11}$  and tetramethyl ether, mp 188°, analysing for  $C_{20}H_{20}O_7$ , indicating the presence of four hydroxyl groups. The UV spectrum had maxima at 320 and 278 nm respectively, indicating it to be an isoflavone. The presence of hydroxyl groups at C-5 and C-7 was indicated by the application of diagnostic shift reagents.

The  $^1H$ NMR spectrum (60 MHz,  $CDCl_3$  + TFA) established its nature as isoflavone. A sharp singlet at  $\delta$  7.99 was assigned to a C-2 proton, a *meta* coupled singlet (1H) at 7.2 due to a C-2' proton and a broad singlet (2H) at 6.97 assigned to C-5' and C-6' protons. The C-8 proton was located at 6.66 which moved downfield to 7.20 on acetylation and methoxyl protons resonated at 4.00. Retro-Diels-Alder ions at  $m/z$  148 and 168 established the presence of one hydroxyl and one methoxyl in ring B and three hydroxyls in ring A.

The positions of the methoxyl group at C-3' and hydroxyl group at C-4' in the ring B were ascertained by the oxidative degradation of 1 with permanganate to give a compound, mp 207°, identified as vanillic acid (lit. mp 210°) [7] (co-TLC, mmp).

Compound 1 on methylation afforded a tetramethyl ether, mp 188°, analysing for  $C_{20}H_{20}O_7$ , and was found to be identical in all respects (co-TLC, mmp,  $^1H$ NMR) with the dimethyl ether of junipegeenin-B, mp 186° [8], thereby confirming the structure of 1.

*Iris kumaonensis* on chemical investigation gave six isoflavones, identified as iriskumaonin, iriskumaonin methyl ether, irisflorentin, junipegeenin-A, irigenin and iridin. All the compounds were identified from their UV, IR,  $^1H$ NMR and mass spectra.

### EXPERIMENTAL

The plant *Iris milesu* was collected from Budhal near Poonch (J & K State, India) and *Iris kumaonensis* from Mari (Lahaul), (Voucher specimen no 12603, Herbarium, RRL, Jammu). Dry rhizomes (1.5 kg) of each species were separately dried, finely powdered and the defatted material was extracted with MeOH (3 × 3 l). The MeOH extracts were concd *in vacuo* to afford ca 150 g of each residue.

The MeOH-residue (20 g) of *Iris milesu* was subjected to CC over silica gel (1.5 kg) and eluted successively with  $CHCl_3$ -MeOH mixtures of increasing polarity. Compound 1 was obtained from  $CHCl_3$ -MeOH (17/3) eluates as yellow needles (40 mg), [ $R_f$  0.65 ( $CHCl_3$ -MeOH, 17/3), 0.50 ( $C_6H_6$ -EtOAc, 2/3)], mp 257°, (Found C, 60.60, H, 3.79, required C, 60.62, H, 3.81%), analysed for  $C_{16}H_{12}O_7$ . UV  $\lambda_{max}^{MeOH}$  nm 278, 320 (sh), +  $AlCl_3$  312, 340 (sh), +  $AlCl_3$  + HCl 300, 345 (sh), + NaOAc 292, 325 (sh), + NaOMe 318, 340 (sh). IR  $\nu_{max}^{KBr}$   $cm^{-1}$  3600–3200 and 1650 (C=O). EIMS (probe) 70 eV,  $m/z$  (rel. int.) [ $M$ ] $^+$  316 (100), 301 (80), 168 (40), 148 (20), 140 (18) and 133 (10). Acetylation of 1 ( $Ac_2O$ - $C_5H_5N$ ) gave a tetraacetate, crystallized from MeOH, needles, mp 88°, analysed for  $C_{24}H_{20}O_{11}$ .  $^1H$ NMR (60 MHz,  $CDCl_3$ )  $\delta$  7.8 (1H, s, H-2), 7.23–7.17 (3H, m, H-8, H-2' and H-5'), 7.03 (1H, br s, H-6'), 3.8 (3H, s, OMe), 2.33, 2.18, 2.17, 2.16 (12H, 4s, 4 × OAc). Methylation of 1 with  $DMS-K_2CO_3$ -Me<sub>2</sub>CO gave a tetramethyl ether, plates from MeOH, mp 188°, analysed for  $C_{20}H_{20}O_7$ .  $^1H$ NMR ( $CDCl_3$ , 90 MHz)  $\delta$  7.76 (1H, s, H-2), 7.12 (1H, br s, H-2'), 6.95 (1H, d, J = 8 Hz, H-6'), 6.80 (1H, d, J = 8 Hz, H-5'), 6.63 (1H, s, H-8), 3.88 (15H, s, 5 × OMe).

The MeOH residue (20 g) of *Iris kumaonensis* on silica gel (1.5 kg) CC with eluents  $CHCl_3$ -*n*-hexane and  $CHCl_3$ -MeOH mixtures in increasing polarity afforded the compounds, iriskumaonin methyl ether from  $CHCl_3$ -*n*-hexane (4/1) eluates, needles (50 mg), mp 183°, analysed for  $C_{19}H_{16}O_7$ , irisflorentin from  $CHCl_3$ -*n*-hexane (9/1) eluates, needles (200 mg), mp 180°, analysed for  $C_{20}H_{18}O_8$ , iriskumaonin from  $CHCl_3$ -MeOH (4/1) eluates, needles (500 mg), mp 207–208°, analysed for  $C_{18}H_{14}O_7$ , irigenin from  $CHCl_3$ -MeOH (19/1) eluates, yellow needles (2 g), [ $R_f$  = 0.6 ( $CHCl_3$ -MeOH, 25/2), 0.7

( $C_6H_6$ -EtOAc, 2:3], mp 185°, analysed for  $C_{18}H_{16}O_8$ , junipeenin-A from  $CHCl_3$ -MeOH (17:3) eluates, analysed for  $C_{16}H_{12}O_7$ , iridin from  $CHCl_3$ -MeOH (4:1) eluates, amorphous powder (2 g), mp 208°, analysed for  $C_{24}H_{26}O_{13}$

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## ISOSOJAGOL, A COUMESTAN FROM *PHASEOLUS COCCINEUS*

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**Key Word Index**—*Phaseolus coccineus*, Leguminosae, coumestan, isosojagol

**Abstract**—A novel coumestan isolated from *Phaseolus coccineus* has been characterized as 3,9-dihydroxy-10-( $\gamma,\gamma$ -dimethylallyl)-coumestan and named isosojagol

#### INTRODUCTION

Previous research has resulted in the isolation of five coumestans from *Phaseolus* species following either infection with fungi or treatment with  $CuCl_2$ . Coumestrol 1 has been found to occur in *P. vulgaris*, *P. lunatus*, *P. aureus* and *P. calcaratus* [1], psoralidin 2 has been detected in *P. lunatus* [1], sojagol 3 has been isolated from *P. aureus* [1], phaseol 4 occurs in *P. aureus* [2] and aureol 5 has been obtained from *P. aureus* [2] and *P. mungo* [Adesanya, O'Neill and Roberts, unpublished]. The present report describes the isolation of three coumestans from the runner bean *P. coccineus* which in addition to coumestrol and aureol produces a novel coumestan, isosojagol 6 after treatment with  $CuCl_2$ .

#### RESULTS AND DISCUSSION

The ethyl acetate extract from  $CuCl_2$ -treated *P. coccineus* seedlings was fractionated on a polyamide column using a chloroform-methanol gradient. Purification of fractions by TLC revealed three fluorescent substances which gave purple colours with Fast Blue Salt B reagent [3]. UV spectroscopy suggested that the three compounds may be coumestans and two of the substances were subsequently identified as coumestrol 1 and aureol 5 by a comparison of their TLC, UV, mass spectral and

$^1H$  NMR characters with authentic standards and literature values [2, 4].

The UV spectrum of the third coumestan contains principal maxima at 206, 246 and 347 nm with the mid-wavelength maximum having a lower intensity than that at 347 nm. Addition of sodium acetate produced a bathochromic shift indicating a free hydroxyl at C-3. The presence of one or more other phenolic functions in the molecule was revealed by further UV spectral shifts upon addition of sodium methoxide. The mass spectrum gave a plausible  $[M]^+$  peak at  $m/z$  336 and a fragmentation pattern similar to those observed for the prenylated coumestans psoralidin [5], sojagol [6] and phaseol [2]. Intense signals at  $m/z$  281 and 280 in the spectrum of the new compound could be attributed to loss of  $C_4H_7$  and  $C_4H_8$  radicals from a prenylated  $[M]^+$  at  $m/z$  336. A minor peak at  $m/z$  253 could result from loss of CO from the ion at  $m/z$  281. Such a transition is typical of coumestans in which removal of the lactonic carbonyl is an important fragmentation.

The  $^1H$  NMR spectrum indicated that the compound possessed a  $\gamma,\gamma$ -dimethylallyl side chain rather than a 2,2-dimethylchromene ring. Signals were also observed for five aromatic protons, three of which show *ortho* coupling, one is *meta* coupled and one shows both *ortho* and *meta* coupling. The two possible structures which can account